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are in the molecule that needs to be shown on the Lewis Structure by using the periodic table. (number of non-bonding electrons + 1/2 number of bonding

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4. chemical bonding and molecular structure |

4.3.2 Bond Angle. It is defined as the angle between the orbitals containing bonding electron pairs around the central atom in a molecule/complex ion.

Electronegativity, hardness and atomic number:

Journal of Advances in Chemistry Sen, K.D.; Jorgensen C.K. Electronegativity; K.D. Chemical Hardness, Structure & Bonding,

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phenyl groups are chemically aromatic and show nearly equal bond lengths a significant contribution is the greater electronegativity of the sp² alpha carbon

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there are scales that were derived starting from experimental data such as bond vibrational J. Mullay in Electronegativity, K.D..Sen and C.K

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